#### **M1.**(a) (i) A **and** 3

accept A **and** 39

1

### anomalous result

independent mark
accept not close to other two volumes **or** correct comparison using
the results
ignore does not fit the pattern

1

# (ii) any **one** from:

- volume of water (used)
   allow amount of water (used)
- time (for water to run through)
   accept rate / speed (at which water runs through)
- temperature
- mass / surface area of pad
   accept amount / size / volume / thickness of pad
- same filter funnel ignore other equipment

1

## (iii) any **one** from:

ignore human error unqualified

- incorrect / volume / amount of water added
- reading / volume / amount of water collected
- some water does not go through the pad allow spillage / poorly placed pad
- not enough time allowed for water to drain through accept rate / speed at which water is added
- pads (from one company) not identical / faulty

# (b) (i) any **two** from:

- it was not the best (at absorbing the water)
   accept correct descriptions of 'not the best' / third best or only better than B
- (needed) to absorb more (water)
   allow not absorbing enough (water)
- to improve their image / sales
   accept (needs) to absorb more (water) than A and C for 2 marks

2

# (ii) any **one** from:

- cost (more)
- use (more) resources
- use (more) energy

  must relate to the company

1

[7]

M2. (a) (i) if (fractional) distillation / hydrogenation mentioned as the method = max 1heat / high temperature / hot / vaporise allow thermal decomposition ignore evaporation do not accept 'burns ? do **not** accept temperature < 100 1 catalyst or silica / alumina / porous pot ignore other named catalyst or steam allow heat (the vapour) to a very high temperature / >800°C for 2 marks 1 (ii) C<sub>2</sub>H<sub>3</sub>Cl ignore attempts to balance equation 1 (iii) single bonds between C - H, C - CI and C - Cdo **not** accept symbols outside the bracket 1 (i) so that the amount of plasticiser / (sample of) PVC is the independent / (b) only variable that affects the bending / flexibility of the samples allow because different sizes would give different results accept because size is a control variable ignore references to reliability / precision etc 1 (ii) to improve the <u>reliability</u> (of the investigation)

accept to calculate a mean

accept to check for anomalous results **or** to check the range of results

ignore accuracy / precision etc

1

(iii) 23

correct answer with or without working = 2 marks
if answer is incorrect

allow 
$$\frac{22+23+24}{3}$$

**or** 21 for **1** mark

2

(iv) (PVC) sample had been stretched / used / tested in first three tests accept higher temperature allow worn **or** become weaker ignore (human) error ignore more flexible / softer ignore intermolecular forces

1

(c) does not bend (easily / much)

ignore non-biodegradable / low maintenance

or it is not flexible or it is rigid

ignore sturdy / stronger / harder

1

[10]

М3.	(a) (i) polyethene / poly(ethene)  accept polythene / polyethylene	_
	(ii) needs heat / energy / high temperature / fuel (for cracking)	1
	ignore other processes	1
	produces carbon dioxide / CO <sub>2</sub>	
	ignore use of CO₂ <b>or</b> 'produces carbon'	1

- (b) any **three** from:
  - use water from local sources **or** water from close to home
  - recycle bottles in the UK / close to home
     accept do not recycle in other countries / Asia
  - (reduction in distance travelled) would reduce CO<sub>2</sub> emitted by transport accept use of transport with low / no carbon dioxide emissions
  - use tap water
  - use glass bottles / waxed cartons / metal bottles
     do not accept 'do not use plastic bottles' without an alternative
     material
  - do not put in landfill **or** recycle <u>more</u>
  - reuse / refill plastic bottles
  - <u>tax</u> imported water / plastic bottles (to offset carbon cost)
  - make more / all plastic bottles in UK
     answers must be about the reduction of carbon cost

[6]

**M4.** (a) not broken down by microorganisms **or** not bio-degradable

accept alternative answers such as: do not rot / corrode / fade / react with atmosphere etc any answers which imply the inertness or non-biodegradability of this plastic accept they don't react, they are 'inert' ignore rusting

do **not** accept weathering

(b) (i) (have a) double bond **or** do not have maximum number of (hydrogen) atoms attached

accept can add / react with hydrogen
accept can take part addition reactions
do not accept it is a double bond
do not accept additional reactions
do not accept has 'spare' / 'free' bond
do not accept alkene alone

(ii) single bond between carbon atoms

all atoms correct + 2 'linking' bonds (linking bonds need not go through bracket)

$$\begin{pmatrix}
H & H \\
| & | \\
C - C \\
| & | \\
C1 & H
\end{pmatrix}_{r}$$

n moved to bottom right of <u>bracket</u> i.e. is below  $\frac{1}{2}$  way on the right first 2 marks are possible for chain structures accept  $[-CHCI-CH_2-]_n$ 

(iii) many molecules **or** many monomers

1

1

1

1

1

joined / bonded / linked **or** form long chain molecules / large molecules **or** to form a long chain polymer

accept many alkenes **or** many (ethene) molecules do **not** accept many ethene alone etc. to form a long polymer is not enough for  $2^{nd}$  mark

1

(iv) no other substances formed  $(A + B \rightarrow C)$ 

allow because double bond breaks so other atoms can add allow one product only do **not** accept saturation occurs

-

[8]

**M5.** (a) (i) by heating

pressure is neutral

using a catalyst/pot/ceramic/porcelain/aluminium oxide

1

(ii) use bromine water/(alkaline) permanganate accept bromine

1

alkene makes bromine go colourless or lose its colour accept alkane does not change the red/orange colour of bromine **not** change colour/goes clear

1

either of these must show bonds at end

1

or

not H on ends

allow 3 instead of n not any other number

(ii) poly(ethene) – brackets not essential accept polythene

1

(iii) **large amount** of waste polymer/poly(ethene)/polythene/litter accept large amount of crude oil **or** finite resource used

1

it is not biodegradable
accept it does not
decompose/decay/break down
it causes pollution/it creates toxic
fumes when burnt are neutral
not it is not recyclable

2

[8]

M6. (a) catalyst 1 (b) (i) made up of **only** carbon and hydrogen 1 (ii)  $C_8H_{18}$ 1 (c) (i) ethene 1 (ii) polymerisation 1 [5]

M7.		(a)	organic	1
		sea	diment	
				1
	(b)	(i)	gases	
				1
		(ii)	bitumen	1
	(c)	(i)	cracking	
			accept thermal decomposition	
			do <b>not</b> accept endothermic	1
		(ii)	many <b>or</b> short <b>or</b> small (ethene) molecules	
		(,,,	accept monomer	
			accept double bonds open up <b>or</b> break	1
			join to make larger molecules	
			accept polymer	
			accept polymerisation	
			accept short chain to long chain ( <b>or</b> molecules) do <b>not</b> accept unsaturated to saturated	1
	,			
	(d)	pod	or ventilation	
			accept limited air supply accept insufficient oxygen	1
		cau	uses incomplete combustion	-
			accept produces CO	1

(fumes contain) carbon monoxide which dangerous toxic is **not** awarded a mark do **not** accept harmful or poisonous

[10]

1

(a)  $C_2H_4$ M8. 1 HHHHCCCHHHHAccept even if in wrong columns 1 polythene or poly(ethene) (c) (i) 1 (ii) addition 1 [4]