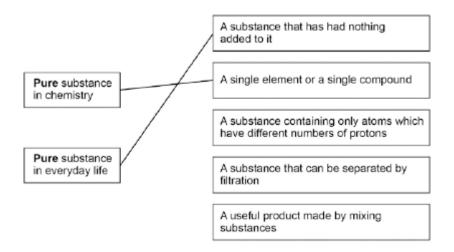
M1.(a) Air

2

Steel

1

(b)



Allow 1 mark for the correct meanings linked to context but incorrect way around

1

1

(c) Damp litmus paper turns white

1

1

(d) Iron(III)

[6]

M2. (a)	(i) so there are no impurities accept no other chemicals / not contaminated allow to get the correct result	1	
	(ii) high melting point	1	
	unreactive	1	
	(iii) yellow-orange	1	
(b)	(i) bubbles / fizz / effervescence ignore any named gas	1	
	(ii) milky	1	
(c)	fast(er)	1	
	small(er) amount	1 [8	3]

//3. (a)	(i)	yellow		1	
			(ii) lilac	1	
		(b)	(bubble through) limewater	1	
			cloudy allow white / milky	1	
		(c)	(i) silver nitrate solution	1	
			(ii) white	1	16

M4.		(a)	(i)	fermentation	1
		(ii)	cloudy accept milky / white	1
				there is carbon dioxide / CO ₂ accept calcium carbonate forms	1
	(b)	(i)		allow a (white) solid / precipitate forms (the amount of ethanol used) increases (from 1970) to 1989 if no year(s) or incorrect year(s) indicated then max 1 correct year(s) only needs to be indicated once to gain full marks accept values in range 1987-1992	1
				then it decreases from 1989 (to 2000)	1
		(ii)	any one from:	
				Brazil had more oilfields	
				cost of crude oil had decreased	
				cost of ethanol / sugar (cane) had increased	
				demand for ethanol / sugar (cane) had increased	
				availability of ethanol / sugar (cane) had decreased accept availability of land to grow sugar (cane) had decreased	
				climate change affects growing sugar (cane)	٠

```
M5.(a)
          (i)
               react
                          allow neutralise
                          allow bubbles / fizzes
                          accept produces gas / CO<sub>2</sub>F
                          ignore rises
                                                                                                          1
                  stop reacting / producing
             (ii)
                         stops on its own is insufficient allow stop working / bubbling /
                         fizzing
                                                                                                          1
                   the (hydrochloric) acid / (calcium) carbonate is used up
                          accept because the (calcium) carbonate has neutralised the
                          (hydrochloric) acid
                   OR
                   have been used up (1)
                   the graph line becomes horizontal / levels out (1)
                   OR
                   stays the same / no change (1)
                          ignore reference to graph line
                   no further reaction (1)
                                                                                                          1
             (iii) bubble the gas through limewater / calcium hydroxide solution
                          allow (add) limewater
                          test must be correct to gain result mark
                                                                                                          1
                   (the solution) goes cloudy
                          allow milky
                                                                                                          1
```

(b) advantage > Quarrying limestone provides building materials, employment and new road links

1

disadvantage > Quarrying limestone releases dust, and lorries release carbon dioxide from burning diesel fuel

1

[7]

M6. (a) (i) milky

carbonate ions

(ii) red

(b) (i) smaller

(iii) The answer obtained is closer to the true value

[5]

M7.		(a)	stop them reacting owtte	1	
	(b)	(i)	fizzing / bubbles / effervescence owtte	1	
		(ii)	(g)	1	
		(iii)	limewater	1	
	(c)	yell	low	1	
	(d)	(i)	barium chloride	1	
		(ii)	white	1	
		(iii)	eg don't see what is being bought ignore references to cost or		
			a comment about quality / purity eg may be impure / contaminated	1	[8]

M8.	(a)	(i)	get wrong coloured flame/result owtte		
		or			
		tog	get the correct result allow contaminated	1	
	(ii)	hig	th melting point	1	
		unr	reactive	1	
	(iii)	ye	llow-orange	1	
(b)	(i)	bul	bbles / fizz / effervescence ignore any named gas	1	
	(ii)	mil		1	
(c)	fast	small(er) amount		1	
	sma			1 [8	3]

##

(a) F

accept indium / In

1

(b) C

accept sodium / Na

1

(c) A

accept hydrogen / H / H₂

1

[3]