

<b>New Document</b>	1	Name.		
		Class:		
		Date:		
		Date.	<del> </del>	
Time:	24 minutes			
Marks:	24 marks			
Comments:				

# Q1.

Sulfur is a non-metal.

Sulfur burns in the air to p	produce sulfur dioxide, S	$SO_2$
------------------------------	---------------------------	--------

(a)	Why is it important that sulfur dioxide is <b>not</b> released into the atmosphere?	
	Tick (✔) one box.	
	Sulfur dioxide causes acid rain.	
	Sulfur dioxide causes global dimming.	
	Sulfur dioxide causes global warming.	
		(1
(b)	Sulfur dioxide dissolves in water.	
	What colour is universal indicator in a solution of sulfur dioxide? Give a reason for your answer.	
		_
		_
		_ (2
(c)	Sulfur dioxide is a gas at room temperature.	
	The bonding in sulfur dioxide is covalent.	
	Explain, in terms of its structure and bonding, why sulfur dioxide has a low boiling point.	
		_
		_
		_ (3

(d) In this question you will be assessed on using good English, organising information clearly and using specialist terms where appropriate.

Sulfur dioxide is produced when fossil fuels are burned.

It is important that sulfur dioxide is not released into the atmosphere.

Three of the methods used to remove sulfur dioxide from gases produced when fossil fuels are burned are:

- wet gas desulfurisation (W)
- dry gas desulfurisation (D)
- seawater gas desulfurisation (S).

Information about the three methods is given in the bar chart and in **Table 1** and **Table 2**.

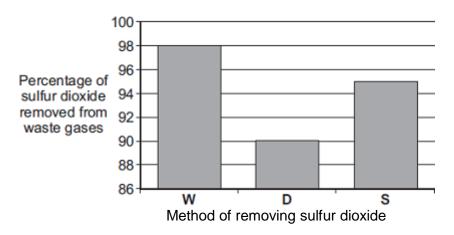


Table 1

Method	Material used	How material is obtained
w	Calcium carbonate, CaCO <sub>3</sub>	Quarrying
D	Calcium oxide, CaO	Thermal decomposition of calcium carbonate:  CaCO <sub>3</sub> — CaO + CO <sub>2</sub>
s	Seawater	From the sea

Table 2

Method	What is done with waste material
W	Solid waste is sold for use in buildings. Carbon dioxide is released into the atmosphere.
D	Solid waste is sent to landfill.
s	Liquid waste is returned to the sea.

Evaluate the three methods of removing sulfur dioxide from waste gases.

Compare the three methods and give a justified conclusion.


(6)

(Total 12 marks)

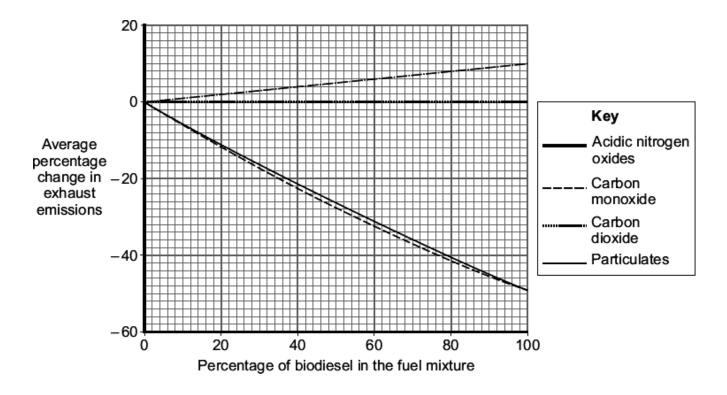
# Q2.

Petroleum diesel is produced from crude oil.

Most vehicles that use petroleum diesel as fuel can also use biodiesel or a mixture of these two fuels. In the UK (in 2010) there must be 5 % biodiesel in all petroleum diesel fuel.

Biodiesel is produced from plant oils such as soya. The crops used to produce biodiesel can also be used to feed humans. The benefit that biodiesel is 'carbon neutral is outweighed by the increasing demand for crops. This increasing demand is causing forests to be burnt to provide land for crops to produce biodiesel. Only a huge fall in the price of petroleum diesel would halt the increasing use of biodiesel.

The graph shows the average percentage change in exhaust emissions from vehicles using different mixtures of petroleum diesel and biodiesel.



There is no difference in carbon dioxide emissions for all mixtures of petroleum diesel and biodiesel.

Use the information and your knowledge and understanding to evaluate the use of plant oils to produce biodiesel.

Remember to give a conclusion to your evaluation.		

# Q3.

Read the article and then answer the questions.

# Supermarkets launch eco-friendly plastic milk bags. Could this be the end of the milk bottle?



Milk bottles are made from glass or from plastic.

Glass milk bottles contain 0.5 litres of milk. When the milk is used up the empty bottles are returned to be re-used. Glass milk bottles are re-used 24 times on average. The glass to make new milk bottles is produced when a mixture of sand, limestone, soda and recycled glass is heated to about 1600 °C in a furnace. There are almost unlimited amounts of the raw materials needed to produce this glass. About 35% of used glass is recycled.

The most common plastic milk bottles contain 2 litres of milk. When the milk is used up the empty bottles are discarded as waste. The plastic used to make these milk bottles is poly(ethene). Poly(ethene) is produced from crude oil by first using fractional distillation, then cracking the naphtha fraction and finally polymerising the ethene. About 5% of used poly(ethene) is recycled.

The new plastic milk bags contain 2 litres of milk. The milk bags are also made from poly(ethene). A milk bag uses 75% less poly(ethene) than is used to make the poly(ethene) milk bottles. When the milk is used up the empty bags are discarded as waste.

(a)	Describe what happens in fractional distillation so that fractions, such as naphtha, are separated from crude oil.

(b)	Supermarkets claim that using milk bags instead of milk bottles would have less environmental impact.
	Do you agree with this claim?
	Use the information in the article and your knowledge and understanding to make appropriate comparisons to justify your answer.
	(Total 7 marks

# Mark schemes

#### Q1.

(a) Sulfur dioxide causes acid rain.

1

(b) red / orange / yellow

do not accept any other colours

1

because sulfur dioxide (when in solution) is an acid

1

(c) (there are) <u>weak</u> forces (of attraction)

do not accept any reference to covalent bonds breaking

1

between the molecules

do not accept any other particles

1

(these) take little energy to overcome

award third mark only if first mark given

1

(d) Marks awarded for this answer will be determined by the Quality of Communication (QC) as well as the standard of the scientific response. Examiners should also refer to the information on page 5 and apply a 'best-fit' approach to the marking.

#### 0 marks

No relevant content

#### **Level 1 (1 – 2 marks)**

A relevant comment is made about the data.

#### Level 2 (3 – 4 marks)

Relevant comparisons have been made, and an attempt made at a conclusion.

#### **Level 3 (5 – 6 marks)**

Relevant, detailed comparisons made and a justified conclusion given.

# examples of the points made in the response

#### effectiveness

- W removes the most sulfur dioxide
- D removes the least sulfur dioxide

#### material used

- Both W and D use calcium carbonate
- Calcium carbonate is obtained by quarrying which will create scars on landscape / destroy habitats
- D requires thermal decomposition, this requires energy
- D produces carbon dioxide which may cause global warming / climate change

S uses sea water, this is readily available / cheap

#### waste materials

- W product can be sold / is useful
- W makes carbon dioxide which may cause global warming / climate change
- D waste fill landfill sites
- S returned to sea / may pollute sea / easy to dispose of

[12]

Q2.

any four from:

to gain 4 marks both pros and cons should be given

# **Arguments for biodiesel**

max three from:

- sustainable / renewable
- (carbon neutral) absorbs CO<sub>2</sub> when growing / during photosynthesis
- burning biodiesel produces low amounts particulates / carbon monoxide allow burning biodiesel produces little / low amount of global dimming ignore sulfur dioxide
- can use waste vegetable oils / fats (from food industry) or can use waste plant material
- can be used to conserve crude oil (instead of / mixed with petroleum diesel)
- produced by a low energy / temperature process
   accept produced by a low tech process
- biodegrades (easily)

  ignore engine effects

#### Arguments against biodiesel

max three from:

- creates food shortages
   accept price of food increases
- deforestation to plant more crops leads to loss of habitat / biodiversity or deforestation leads to a reduction in absorption of CO<sub>2</sub>

allow burning trees increases CO<sub>2</sub> allow deforestation increases global warming

 burning biodiesel produces high amounts of nitrogen oxides allow increases acid rain

- crops takes time to grow allow crops can fail
- vast areas of land needed to grow crops

conclusion supported by the argument presented, which must give added value to the points for and against given above

1

[5]

4

# **Q3.**

(a) allow answers referring specifically to the naphtha fraction

crude oil is evaporated/vaporised (by heating)

1

the vapours are <u>condensed</u> (by cooling)

1

(fractions condense) / boil at different temperatures allow fractions have different boiling points

1

(b) any four from:

answer yes or no does not gain credit ignore references to volume of milk held / number of bottles used / biodegradability / habitats / pollution / mining / dust each marking point must be a comparison

#### milk bag points

- uses (75%) less crude oil to make (than a plastic milk bottle)
   allow eg uses 75% less
   poly(ethene) which is made from crude oil
- uses less energy / fuel to make (than a plastic / glass milk bottle)
- produces less carbon dioxide to manufacture (than a plastic / glass milk bottle)

allow produces less greenhouse gases / causes less global warming

allow produces less CO<sub>2</sub> on burning

- produces less waste (than a plastic / glass milk bottle)
   allow takes up less landfill (space)
   allow an argued case for more waste eg milk bags are
   discarded / cannot be reused
- less fuel used for transport than glass milk bottles
- (produces waste because) milk bags are only used once whereas glass bottles can be re-used

allow milk bags are discarded but glass bottles can be reused (24 / many times)

allow glass bottles can be reused but milk bags can't

# poly(ethene) points

- uses a limited raw material / crude oil whereas the raw materials for glass are almost unlimited
- less (5%) poly(ethene) is recycled (compared to glass (35%))

  allow (35%) glass is recycled or (5%) poly(ethene) (bottles)
  recycled BUT milk bags aren't / are discarded

OI

recycled poly(ethene) is not used to make new bags whereas recycled glass is used to make new bottles

4